




Unit 7 – Logarithmic Functions  
Assignment 1 – Unit Assignment

STUDENT NAME / QUIZ				ANSWERS											
Answer Key				1	A	B	C	D	E	16	A	B	C	D	E
STUDENT ID				2	A	B	C	D	E	17	A	B	C	D	E
0	0	0	0	3	A	B	C	D	E	18	A	B	C	D	E
1	1	1	1	4	A	B	C	D	E	19	A	B	C	D	E
2	2	2	2	5	A	B	C	D	E	20	A	B	C	D	E
3	3	3	3	6	A	B	C	D	E	21	A	B	C	D	E
4	4	4	4	7	A	B	C	D	E	22	A	B	C	D	E
5	5	5	5	8	A	B	C	D	E	23	A	B	C	D	E
6	6	6	6	9	A	B	C	D	E	24	A	B	C	D	E
7	7	7	7	10	A	B	C	D	E	25	A	B	C	D	E
8	8	8	8	11	A	B	C	D	E	26	A	B	C	D	E
9	9	9	9	12	A	B	C	D	E	27	A	B	C	D	E
 <b>QuickKey</b> get.quickkeyapp.com				13	A	B	C	D	E	28	A	B	C	D	E
 <b>QuickKey</b> get.quickkeyapp.com				14	A	B	C	D	E	29	A	B	C	D	E
 <b>QuickKey</b> get.quickkeyapp.com				15	A	B	C	D	E	30	A	B	C	D	E

**Part 1 – Selected Response:**

**Instructions:** Choose the best answer and shade it in the corresponding space on the answer sheet provided.

\_\_\_\_\_ 1. What is the x-intercept of  $f(x) = -3 \ln x$ ?

- (A) -1
- (B) 0
- (C) 1
- (D) -3

\_\_\_\_\_ 2. What is the end behaviour of  $f(x) = 5 \log x$ ?

- (A) I to II
- (B) I to IV
- (C) II to I
- (D) IV to I

inc.  
I

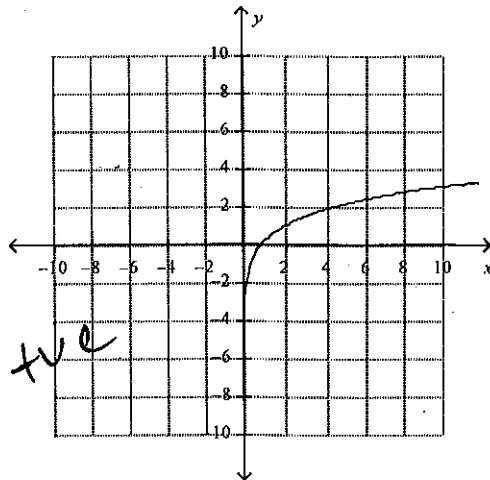
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IV

3. What is the function that gives the graph provided?

- (A)  $f(x) = -\frac{1}{3} \ln x$
- (B)  $f(x) = 3 \log x$
- (C)  $f(x) = -\frac{1}{3} (3)^x$
- (D)  $f(x) = 0.3(10)^x$

inc  
log a +ve



4. Which of the following describes the function  $f(x) = 2 \log x$ ?

- (A) Decreasing with a x-intercept of 1
- (B) Decreasing with a x-intercept of 2
- (C) Increasing with a x-intercept of 1
- (D) Increasing with a x intercept of 2

inc x int @ 1

5. What is the exponential form of  $y = \log_4 \frac{20}{15}$ ?

- (A)  $20^y = 4$
- (B)  $4^y = 20$
- (C)  $y^{20} = 4$
- (D)  $y^4 = 20$

$4^y = 20$

6. What is an estimate the value of  $y$  in the exponential equation  $50 = e^y$ ?

- (A) 0.3
- (B) 1.7
- (C) 2.9
- (D) 3.9

$\ln 50 = y$

7. Determine the concentration of hydrogen ions in bleach, with a pH of 12.8. Recall that pH,  $p(x)$ , is defined by the equation  $p(x) = -\log x$  where the concentration of hydrogen ions,  $x$ , in a solution is measured in moles per liter.

- (A)  $1.3 \times 10^{-13} \text{ mol/L}$
- (B)  $1.6 \times 10^{-13} \text{ mol/L}$
- (C)  $1.3 \times 10^{-12} \text{ mol/L}$
- (D)  $1.6 \times 10^{-12} \text{ mol/L}$

con =  $10^{-\text{pH}}$

8. What is  $\log_3 100 - 2\log_3 5$  written as a single logarithm?

- (A)  $\log_3 4$
- (B)  $\log_3 10$
- (C)  $\log_3 1000$
- (D)  $\log_3 2500$

$\log_3(100) - \log_3(5^2)$   
 $\log_3\left(\frac{100}{25}\right)$

9. Evaluate:  $\log_2\left(\frac{1}{128}\right)$

(A) -64  
 (B) -7  
 (C) 7  
 (D) 64

$\frac{\log\left(\frac{1}{128}\right)}{\log 2} =$

10. What is the logarithmic form of  $C = 2^m$ ?

(A)  $m = \log_2 C$   
 (B)  $m = \log_m 2$   
 (C)  $C = \log_2 m$   
 (D)  $C = \log_m 2$

$\log_2(C) = m$

11. Evaluate:  $3\log_2 8 - 3\log_2 4$

(A) 2  
 (B) 3  
 (C) 4  
 (D) 18

$3\left(\frac{\log 8}{\log 2}\right) - 3\left(\frac{\log 4}{\log 2}\right)$   
 $3(3) - 3(2)$   
 $9 - 6 =$

12. Which value is the best estimate for  $y = \log_4 500$ ?

(A) 0.2  
 (B) 0.6  
 (C) 2.7  
 (D) 4.5

$\frac{\log 500}{\log 4}$

13. Solve:  $4^{x+1} = 7$

(A)  $\frac{\log 4}{\log 7} - 1$   
 (B)  $\frac{\log 7}{\log 4} - 1$   
 (C)  $\frac{\log 4 - 1}{\log 7}$   
 (D)  $\frac{\log 7 - 1}{\log 4}$

$\log_4(x) = x + 1$   
 $\frac{\log 7}{\log 4} - 1 = x + 1$   
 $\frac{\log 7}{\log 4} - 1 = x$

14. The number of ants in a colony is modelled by the equation  $N(d) = 120(1.14)^w$  where  $N(d)$  represents the number of ~~bees~~ <sup>ants</sup> and  $w$  represents the number of weeks from now. After how many ~~days~~ <sup>weeks</sup> will there be 200 ants?

- (A) 4  
(B) 5  
(C) 6  
(D) 7

$$\frac{200}{120} = \frac{120}{120} (1.14)^w$$

$$1.67 = (1.14)^w$$

$$\frac{\log 1.67}{\log 1.14} = w \frac{\log(1.14)}{\log 1.14}$$

15. Which expression is equivalent to  $\log\left(\frac{\sqrt{AB}}{C^3}\right)$ ?

- (A)  $\frac{1}{2}\log A - \log B + 3\log C$   
(B)  $\frac{1}{2}\log A + \log B - 3\log C$   
(C)  $2\log A - \log B + \frac{1}{2}\log C$   
(D)  $2\log A + \log B - \frac{1}{3}\log C$

$$\log \frac{A^{\frac{1}{2}} B}{C^3}$$

$$\frac{1}{2} \log A + \log B - 3 \log C$$

### Part 2 – Constructed Response:

**Instructions:** Complete all of the following in the space provided. For full marks be sure to show all workings and present your answers in a clear and concise manner.

1. Simplify then evaluate:

(A)  $4\log_4 2 - \log_4 8$  (1)

$$= \log_4 (2^4) - \log_4 (8)$$

$$= \log_4 (16) - \log_4 (8)$$

$$= \log_4 \left(\frac{16}{8}\right)$$

$$= \log_4 (2)$$

$$= \frac{\log 2}{\log 4} = \frac{1}{2}$$

(B)  $\frac{1}{2}\log 64 + 3\log 5$  (1)

$$= \log (64^{\frac{1}{2}}) + \log (5^3)$$

$$= \log (8) + \log (125)$$

$$= \log (8 \cdot 125)$$

$$= \log (1000)$$

$$= \log 10^3 = 3$$

2. Algebraically solve:  $6^{x-2} = 5^{x+3}$  (2)

$$(x-2) \log 6 = (x+3) \log 5$$

$$x \log 6 - 2 \log 6 = x \log 5 + 3 \log 5$$

$$x \log 6 - x \log 5 = 3 \log 5 + 2 \log 6$$

$$\frac{0.079x}{0.079} = \frac{3.653}{0.079} \quad \therefore \quad x = 46.24$$

3. After taking a cough suppressant, the amount, A, in mg, remaining in the body is given by:

$$A = 400 \left(\frac{1}{2}\right)^t, \text{ where } t \text{ is given in hours.}$$

(A) What is the initial amount taken? (1)

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400 mg

(B) What percent of the drug leaves the body each hour? (1)

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50%

(C) Algebraically determine how much of the drug is left in the body 6 hours after the dose is given? (1)

$$A = 400 \left(\frac{1}{2}\right)^6 = 6.25 \text{ mg}$$

(D) How long is it until only 1 mg of the drug remains in the body? (2)

$$\frac{1}{400} = 400 \left(\frac{1}{2}\right)^t \quad t = 8.6 \text{ h.}$$

$$\frac{\log \left(\frac{1}{400}\right)}{\log \left(\frac{1}{2}\right)} = t$$

4. The pH scale is used to measure the acidity of a solution. The pH,  $p(x)$ , is defined by the equation  $p(x) = -\log x$ , where the concentration of hydrogen ions,  $x$ , in a solution is measured in moles per litre (mol/L).

- (A) Black coffee has a pH of 5. What is its hydrogen ion concentration? (1)

$$\begin{aligned} x &= 10^{-p(x)} \\ &= 10^{-5} \text{ mol/L} \end{aligned}$$

- (B) Water has a pH of 7. In terms of concentration, how much more acidic is black coffee than water? (2)

$$\begin{aligned} x &= 10^{-pH} \\ &= 10^{-7} \text{ mol/L} \end{aligned}$$

con. coffee  $10^{-5}$  mol/L  
con. water  $10^{-7}$  mol/L

$$\frac{10^{-5}}{10^{-7}} = 10^2 = 100$$

coffee is 100 times more acidic.

5. Identify any error(s) in the solution below and provide the correct solution. (3)

$$\begin{aligned} &\frac{1}{2} \log_2 36 + (2 \log_2 6 - \frac{1}{2} \log_2 81) \\ &= \log_2 36^{\frac{1}{2}} + (\log_2 6^2 - \log_2 81^{\frac{1}{2}}) \\ &= \log_2 18 + (\log_2 12 - \log_2 9) \\ &= \log_2 18 + (\log_2 3) \\ &= \log_2 21 \end{aligned}$$

①  $\log_2(36^{\frac{1}{2}})$  is  $\log_2 \sqrt{36}$  not

$36/2$ , should be  $\log 6$

②  $\log_2 6^2$  is  $\log_2 (6 \cdot 6) = \log 36$   
not  $\log (6 \cdot 2)$ .

③ when sub. logs you divide arguments.  $\log_2(12) - \log_2 9 = \log(\frac{12}{9})$

4) when adding logs you multiply arguments,  $\log 18 + \log 3$   
should be  $\log(18 \cdot 3)$

$$\begin{aligned} \cdot 5) \text{ cont} & \quad \frac{1}{2} \log_2(36) + (2 \log_2 6 - \frac{1}{2} \log_2 81) \\ & = \log_2(36^{\frac{1}{2}}) + [\log_2(6^2) - \log_2(81^{\frac{1}{2}})] \\ & = \log_2 6 + [\log_2 36 - \log_2 9] \\ & = \log_2 6 + \log_2 \left( \frac{36}{9} \right) \\ & = \log_2 6 + \log_2 4 \\ & = \log_2(6 \cdot 4) \\ & = \log_2(24) \end{aligned}$$