## Part 1: Circle the letter of the correct answer.

1. What is the end behavior of $y=\frac{1}{3} \ln x$ ?
(A) I to II
(B) I to IV
(C) II to I
(D) IV to I
2. What is the end behavior of $y=-\frac{1}{2} \log x$ ?
(A) I to II
(B) I to IV
(C) II to I
(D) IV to I
3. What is the $x$-intercept of $y=-2 \log x$
(A) $(1,0)$
(B) $(-2,0)$
(C) $(0,1)$
(D) $(0,2)$
4. What is the domain of $y=-\frac{1}{4} \log x$ ?
(A) $\{x \mid x>0, x \varepsilon R\}$
(B) $\{x \mid x \geq 0, x \varepsilon R\}$
(C) $\{x \mid x<0, x \varepsilon R\}$
(D) $\{x \mid x \leq 0, x \varepsilon R\}$
5. What is $\log \mathrm{A}+3 \log \mathrm{~B}-5 \log \mathrm{C}$ written as a single logarithm?
(A) $\frac{4(\log \mathrm{~A}+\log \mathrm{B})}{2 \log \mathrm{C}}$
(B) $\quad \log \left(\mathrm{A}^{4}+\mathrm{B}-\mathrm{C}^{2}\right)$
(C) $\quad \log \left(\frac{\mathrm{A} \cdot \mathrm{C}}{\mathrm{C}^{2}}\right)$
(D) $\quad \log \frac{\mathrm{A} \cdot \mathrm{B}^{3}}{\mathrm{C}^{5}}$
6. Which is equivalent to $\log _{4}\left(\frac{A B^{3}}{C}\right)$ ?
(A) $\quad \log _{4}\left(A+B^{3}-C\right)$
(B) $\log _{4} A+3 \log _{4} B-\log _{4} C$
(C) $3\left(\log _{4} A+\log _{4} B\right)-\log _{4} C$
(D) $3\left(\log _{4} A+\log _{4} B-\log _{4} C\right)$
7. Which graph best represents the function $y=-\log x$
(A)

(B)

(C)

(D)

8. What is the logarithmic form of $5^{x}=6$ ?
(A) $\quad \log _{5} x=6$
(B) $\quad \log _{5} 6=x$
(C) $\quad \log _{6} x=5$
(D) $\quad \log _{x} 5=6$
9. What is the exponential form of $\log _{3} 5=x$ ?
(A) $3^{5}=x$
(B) $x^{3}=5$
(C) $\quad 5^{x}=3$
(D) $\quad 3^{x}=5$
10. Evaluate: $\log _{4} 10$.
(A) 0.60
(B) 1.66
(C) 1.78
(D) 2.50
11. Given $5^{x}=12$, which best approximates $x$ ?
(A) 0.65
(B) 1.23
(C) 1.46
(D) 1.54
12. Which graph best represents the function $y=7 \ln x$ ?
(A)

(B)

(C)

(D)

13. Evaluate $3 \log _{8} 24-3 \log _{8} 3$
(A) 3
(B) 7
(C) 1
(D) 64
14. Calculate the pH of a solution with the hydrogen ion concentration of $0.0000065 \mathrm{~mol} / \mathrm{l}$. Recall that $\mathrm{pH}, \mathrm{p}(\mathrm{x})$ is defined by the equation $\mathrm{p}(\mathrm{x})=-\log \mathrm{x}$, where the concentration of hydrogen ions, in a solution is measured in moles per litre.
(A) 6.5
(B) -5.2
(C) -6.5
(D) 5.2
15. What is the range of $y=-2 \log x$
(A) $\{y \mid y>0, \mathrm{y} \varepsilon R\}$
(B) $\{y \mid \mathrm{y} \varepsilon R\}$
(C) $\{y \mid y<0, \mathrm{y} \varepsilon R\}$
(D) $\{y \mid y>-2, \mathrm{y} \varepsilon R\}$

## Part 2: Answer all of the following questions showing all work

16. Evaluate the following using the Laws of Logarithms:
(7)
A) $2 \log _{12} 6+\log _{12} 4$
B) $\quad \frac{1}{2} \log _{2} 36+\log _{2} 5-\log _{2} 15$
17. Algebraically, solve for x :
A) $\quad \log _{4} x=-3$
B) $\quad 7^{x-2}=310$
(6)
18. The half-life of a certain drug in the bloodstream is 4 days. If a patient is given 500 mg , algebraically determine how long it will take for the amount of drug in the patient's body
(4) to reduce to $15 \mathrm{mg} . \quad A=A_{o}\left(\frac{1}{2}\right)^{\frac{t}{h}}$
19. $\$ 2500$ is invested at $7.5 \%$ per year, compounded monthly. How many years will it take for the initial investment to reach $\$ 5000$. Use the following formula: $\quad A=P(1+i)^{n}$
(4)
20. In terms of hydrogen ion concentration, how much more acidic is lemon juice, with a pH of 2, than baking soda, with a pH of $9 ? \quad p(x)=-\log x$
(4)
